

What do you need to make a hot dog?

Can we make an equation for the ingredients for a hot dog?

What do you need to make a hot dog?

Bun
hot dog

Can we make an equation for the ingredients for a hot dog?



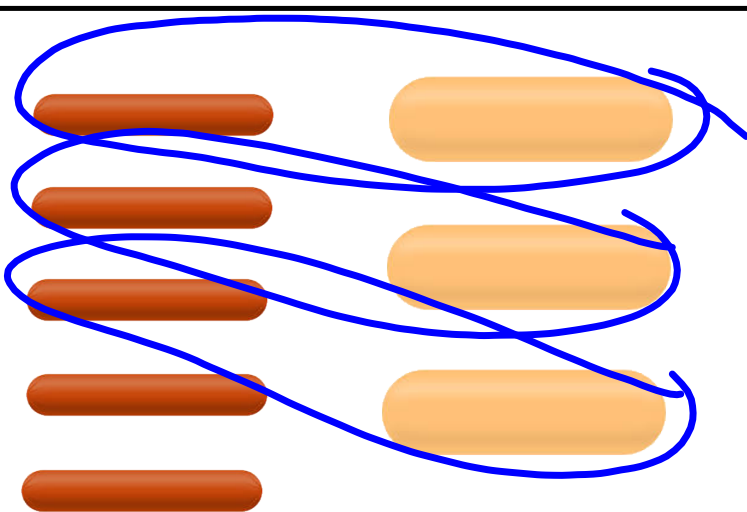
How many hot dogs can I make with the following ingredients?



What is limiting the number of hot dogs I can make?

What ingredient do I have extra of?

How many hot dogs can I make with the following ingredients?



What is limiting the number of hot dogs I can make?

Buns

What ingredient do I have extra of?

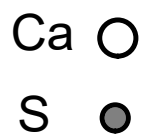
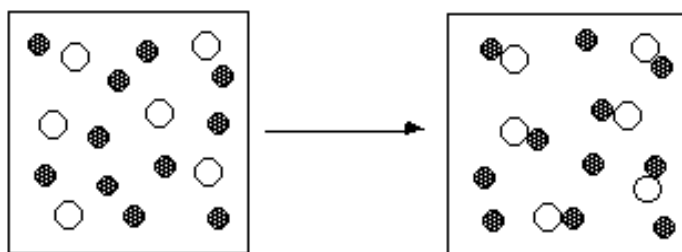
hot dogs

Limiting Reactants

Limiting reactant: the reactant that limits how much product can be produced

Excess reactant: the reactant that will have extra when the reaction is complete

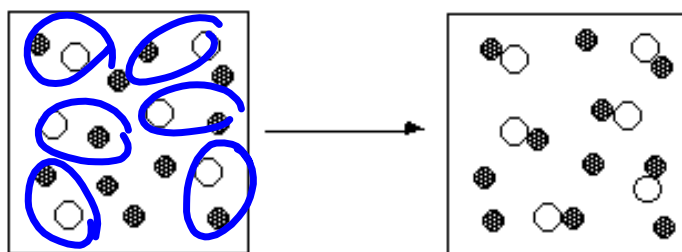
Limiting Reactants:



What is the limiting reactant?

What is in excess?

Limiting Reactants:



Ca ○

S ●

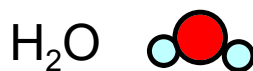
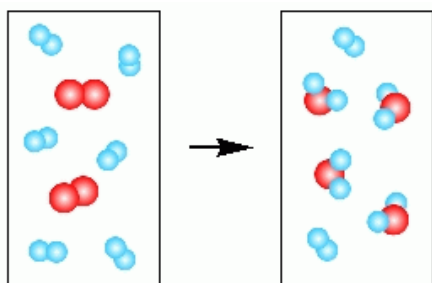


What is the limiting reactant? Ca

What is in excess? S

5 mol S

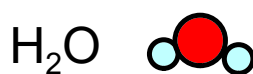
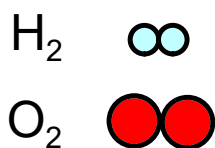
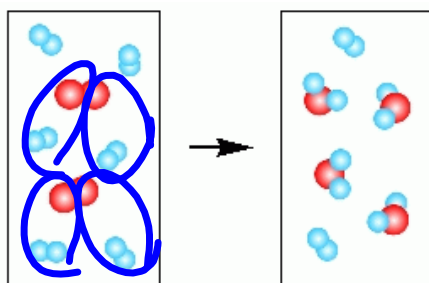
Limiting Reactants:



What is the limiting reactant?

What is in excess?

Limiting Reactants:

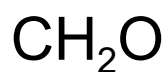
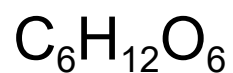


What is the limiting reactant? O₂

What is in excess? H₂ 2 mol H₂

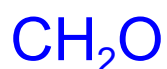
Empirical and Molecular Formulas

What's the difference between the following formulas?



Empirical and Molecular Formulas

empirical formulas: elements are in the lowest ratio



molecular formulas: elements are not in their lowest ratio



Practice

Determine which of the following are empirical formulas and which are molecular formulas. If it is in its molecular formula, change it to its empirical formula:

1. H_2O_2
2. CaCl_2
3. WO_2
4. $\text{C}_2\text{H}_6\text{O}_2$

Practice

Determine which of the following are empirical formulas and which are molecular formulas. If it is in its molecular formula, change it to its empirical formula:

1. H_2O_2 molec. HO
2. CaCl_2 emp.
3. WO_2 emp.
4. $\text{C}_2\text{H}_6\text{O}_2$ molec. CH_3O